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CHAPTER 7 REVIEW Chemical

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SECTION 1 SHORT ANSWER Answer

the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms

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of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical  
Compounds

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## Chemical Formulas And

### Chemical Compounds.. B

#### ANSWERS

#### WRITING CHEMICAL FORMULAS

PRACTICE • Write the formula of the binary ionic compound that will result between the following elements: • 1. zinc and sulfur • 2. aluminum and phosphorus • 3. strontium and chlorine • 4. barium and oxygen •

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Chapter 7: Chemical Formulas and  
Chemical compounds

Chapter Seven: Chemical Compounds  
and Chemical Formulas. Chapter  
interactives: Common chemicals and  
their common names. Naming ionic

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## Chemical Formulas And

and covalent compounds. Oxidation numbers. Oxidation numbers #2. Valence electrons and oxidation state. Oxidation numbers. Molecular and empirical formulas.

Chapter Seven [Chemical Formulas and Chemical Compounds]

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Chapter 7 Notes – Chemistry;  
Chemical Formulas and Chemical  
Compounds Molecules and Molecular  
Compounds • A molecule consists of  
two or more atoms bounded together.  
Molecules and Chemical Formulas •  
Each molecule has a chemical  
formula. • The chemical formula

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Compounds: 1. Which atoms are found in the molecule, and 2.

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Chemical Formulas and ...  
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Formulas. Monatomic ion. Polyatomic  
ion. Anion. Cation. A single atom with

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Answers

a positive or negative charge resulting from.... A tightly bound group of atoms that behaves as a unit and has.... Any atom or group of atoms with a negative charge. Any atom or group of atoms with a positive charge.

formulas chapter 7 chemical names



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Chemical Compounds Chemists use  
chemical names and formulas to  
describe the atomic composition of  
compounds Section 1 Chemical  
Names and Formulas There are  
millions of synthetic and natural

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Chemical compounds many of which still go by their common name or i.e. sodium bicarbonate (baking soda), calcium carbonate (limestone), sodium chloride (table salt) Common names however give you no information about chemical composition Systematic methods were put in place

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Chapter 7: Chemical Formulas and  
Chemical Compounds Section 7-1:

Chemical Names and Formulas 7-1-1

Explain the significance of a chemical

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formula. The subscript indicates that there are 8 carbon atoms in the molecule of octane. The subscript indicates that there are 18 hydrogen atoms in the molecule of octane. Subscript 2 refers to 2 aluminum ...

Chapter 7: Chemical Formulas and

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include: Test for Fe(II) and Fe(III)  
Water of Hydration. Determining the

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chapter 7.1: chemical names and  
formulas. chemical formula.  
Hydrocarbons. monatomic ion. name

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the two types of ions. compound-  
combination of different atoms in a  
fixed ratio. compound made of carbon  
and hydrogen atoms. ions made of  
single atom. cation, anion.

chemical formulas chapter 7 1  
Flashcards and Study Sets ...

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"A chemical formula indicates the \_\_\_\_\_ number of \_\_\_\_\_ of each kind in a chemical compound.", What do subscripts tell you about a chemical compound?, The polyatomic ion chlorite ( $\text{ClO}_2^-$ )- bonds with copper to form  $\text{Cu}(\text{ClO}_2)_2$ , a compound called copper (II) chlorite. How many atoms



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of each element are there in copper chlorite? Write down the name of the following ion:  $V^{4+}$ , and determine whether ...

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Compounds

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Chemistry Pages 218-249 Chapter

7-1: Chemical Names and Formulas

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slide show) on PowerShow.com - id:

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Chemical Formulas & Chemical  
Compounds. Chapter 7 is the final  
chapter of the first semester. It serves  
as a summary of all course content  
discussed so far, as

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