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the measure of how much energy is required to raise 1kg of a s.... to calculate the amount of energy required to heat an object b.... (DELta) $Q = M \times C \times (\Delta)T$. = the difference between. what is specific heat capacity. the measure of how much energy is required to raise 1kg of a s.... how can we use S.H.C.

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a) Temperature is the degree of hotness of a body. b) The upper fixed point on the Celsius scale is the temperature of steam from pure boiling water at standard atmospheric pressure. 2. Liquid-in-glass thermometer— the volume of liquid increases as the temperature increases.

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Answer: – The amount of heat received by an object = mass of the object x relative to the material element x increase the heat. Question: – What are the constants of the Celsius scale of warmth? Answer: – Celsius scale 0oC and upper-temperature 100oC on the Celsius scale of warmth.

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Heat, Temperature, and Thermal Energy • Thermal energy E_{th} is an energy of the system due to the motion of its atoms and molecules. Any system has a thermal energy even if it is isolated and not interacting with its environment. The units of E_{th} are Joules. • Heat Q is energy transferred between the system and

~~Chapter 17. Work, Heat, and the First Law of Thermodynamics~~

Heat is a form of energy that flows from a body of higher temperature to one of lower temperature. Temperature is a degree of hotness or coldness of a body. 2.

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The answer is (c) copper is a better conductor of heat than the stainless steel. Introduction to NCERT Solutions Class 7 Science Chapter 4 Heat This chapter focuses on questions regarding heat and measurement of heat. We will see questions on the difference between clinical and laboratory thermometer.

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