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3 Exercise 8-1:

Using the

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predict and

balance the

following single

replacement

reactions. Use

abbreviations to

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indicate the appropriate phase of reactants and products where possible. Note: Not all of the reactions will occur. For those that do not, write no reaction. 1.

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Determine the

oxidation number

of the

underlined

element: KMnO_4

Since K is an

alkali metal,

its charge must

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be $1+$ Oxygen is
 $2-$ but there are
four of them,
therefore, 4
times $2-$ equals
 $8-$ If $1+$ and $8-$
are added

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many of the

transuranium

elements were

prepared at the

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California at
Berkeley,
California,
U.S.A., it is
logical to find
the following
names:

Americium—named
after the
Americas (after
all, one must be
patriotic!).

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